XAFS Characterization of Industrial Catalysts: Cu-ZnO_x/SiO₂ Catalyst for ester hydrogenolysis

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Introduction

Zinc-doped Cu/SiO2 catalyst is primarily chosen for carbonyl hydrogenation in petrochemical and coal-based chemical industry. The valence states of copper and zinc, the local structure of alloyed Cu-ZnOx (or perhaps not alloyed), the stability of active center, -- all these are still controversial and challengeable for traditional laboratory due to the limit for nanosized structure and in situ characterization. [1,2]

In this investigation, the valence states and local structure of Cu-Zn/SiO₂ and Cu/SiO₂ catalyst during hydrogen reduction at temperatures in the range of 20°C to 350°C has been studied by *in situ* X-ray absorption spectroscopy characterization. In combination with laboratory characterization such as HAADF-STEM, EDX mapping, FTIR, and XRD etc., a double-layer nanoraft structure of Cu-ZnOx was proposed during the reduction of mixed copper-zinc phyllosilicate.

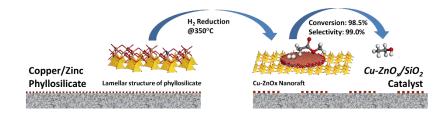


Figure 1 formation of double-layer nanoraft from mixed copper/zinc phyllosilicate.

References

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